

- Q. Write the electronic configuration of the following elements:
Sc, Mn, Fe, Cu and Zn. [The atomic mass are Sc = 21, MN = 25, Fe = 26,
Cu = 29 and Zn = 30.]
- Q. What are transition elements?
- Q. Cu in ground state contains 10 electrons in 3d subshell but it is known as transition element. Explain.
- Q. What are d-Block elements?
- Q. Zn, Cd and Hg not considered as transition elements .Why?